



## **General Certificate of Education**

# **Physics 5451**

## *Specification A*

**PA01      Particles, Radiation and Quantum  
Phenomena**

# **Mark Scheme**

*2009 examination - January series*

Mark schemes are prepared by the Principal Examiner and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation meeting attended by all examiners and is the scheme which was used by them in this examination. The standardisation meeting ensures that the mark scheme covers the candidates' responses to questions and that every examiner understands and applies it in the same correct way. As preparation for the standardisation meeting each examiner analyses a number of candidates' scripts: alternative answers not already covered by the mark scheme are discussed at the meeting and legislated for. If, after this meeting, examiners encounter unusual answers which have not been discussed at the meeting they are required to refer these to the Principal Examiner.

It must be stressed that a mark scheme is a working document, in many cases further developed and expanded on the basis of candidates' reactions to a particular paper. Assumptions about future mark schemes on the basis of one year's document should be avoided; whilst the guiding principles of assessment remain constant, details will change, depending on the content of a particular examination paper.

Further copies of this Mark Scheme are available to download from the AQA Website: [www.aqa.org.uk](http://www.aqa.org.uk)

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## Instructions to Examiners

- 1** Give due credit to alternative treatments which are correct. Give marks for what is correct; do not deduct marks because the attempt falls short of some ideal answer. Where marks are to be deducted for particular errors specific instructions are given in the marking scheme.
- 2** Do not deduct marks for poor written communication. Refer the script to the Awards meeting if poor presentation forbids a proper assessment. In each paper candidates may be awarded up to two marks for the Quality of Written Communication in cases of required explanation or description. Use the following criteria to award marks:
  - 2 marks:** Candidates write legibly with accurate spelling, grammar and punctuation; the answer containing information that bears some relevance to the question and being organised clearly and coherently. The vocabulary should be appropriate to the topic being examined.
  - 1 mark:** Candidates write with reasonably accurate spelling, grammar and punctuation; the answer containing some information that bears some relevance to the question and being reasonably well organised. Some of the vocabulary should be appropriate to the topic being examined.
  - 0 marks:** Candidates who fail to reach the threshold for the award of one mark.
- 3** An arithmetical error in an answer should be marked AE thus causing the candidate to lose one mark. The candidate's incorrect value should be carried through all subsequent calculations for the question and, if there are no subsequent errors, the candidate can score all remaining marks (indicated by ticks). These subsequent ticks should be marked CE (consequential error).
- 4** With regard to incorrect use of significant figures, normally two, three or four significant figures will be acceptable. Exceptions to this rule occur if the data in the question is given to, for example, five significant figures as in values of wavelength or frequency in questions dealing with the Doppler effect, or in atomic data. In these cases up to two further significant figures will be acceptable. The maximum penalty for an error in significant figures is **one mark per paper**. When the penalty is imposed, indicate the error in the script by SF and, in addition, write SF opposite the mark for that question on the front cover of the paper to obviate imposing the penalty more than once per paper.
- 5** No penalties should be imposed for incorrect or omitted units at intermediate stages in a calculation or which are contained in brackets in the marking scheme. Penalties for unit errors (incorrect or omitted units) are imposed only at the stage when the final answer to a calculation is considered. The maximum penalty is **one mark per question**.
- 6** All other procedures, including the entering of marks, transferring marks to the front cover and referrals of scripts (other than those mentioned above) will be clarified at the standardising meeting of examiners.

## GCE Physics, Specification A, PA01, Particles, Radiation and Quantum Phenomena

Question 1		
(a)	same number of protons ✓ different number of neutrons ✓	2
(b)	40 ✓	1
(c)	$19 \times 1.6 \times 10^{-19} \checkmark / 40 \times 1.67 \times 10^{-27} \checkmark$ $= 4.55 \times 10^7 \text{ C kg}^{-1} \checkmark$	3
(d)	$x = 40$ $y = 18 \checkmark$	1
<b>Total</b>		<b>7</b>

Question 2		
(a)	mercury atoms are excited ✓ by electron impact/an electric current ✓ they then emit light upon relaxation ✓ which is (predominantly) ultra violet ✓	4
(b)	the coating absorbs the (ultra violet) light (from the mercury atoms) ✓ and emits light longer wavelength/lower frequency ✓ in the visible region ✓ having a larger range of photon energies (or emits photons in many stages or electrons cascade down energy levels) ✓	max 3
<b>Total</b>		<b>7</b>

Question 3		
(a)	decreasing the wavelength increases the frequency or reference to $f = c/\lambda \checkmark$ the energy of the photon increases with frequency (or reference to $E = hf \checkmark$ ) electrons need a minimum amount of energy to leave the metal ✓ called the work function ✓	max 3
(b)	(i) $f = c/\lambda = 3.00 \times 10^8 / 4.90 \times 10^{-7} = 6.12 \times 10^{14} \text{ Hz} \checkmark$ (ii) $E = hf = 6.63 \times 10^{-34} \times 6.12 \times 10^{14} = 4.06 \times 10^{-19} \text{ (J)} \checkmark$ (iii) $E_k = hf - \phi \checkmark$ $= 4.06 \times 10^{-19} - 3.94 \times 10^{-19} = 1.2 \times 10^{-20} \text{ (J)} \checkmark$ (iv) $1.2 \times 10^{-20} / 1.6 \times 10^{-19} = 0.075 \text{ (eV)} \checkmark$ allow CE	5
<b>Total</b>		<b>8</b>

Question 4			
(a)	(i)	$n = \sin \theta_a / \sin \theta_g$ $1.60 = \sin \theta_a / \sin 36.0 \checkmark$ $\theta_a = 70.1^\circ \checkmark$	5
	(ii)	$\sin \theta = 1/n$ $\theta_c = 38.7^\circ \checkmark$	
	(iii)	ray reflected from the side of the glass block $\checkmark$ ray entering the base without refraction $\checkmark$ (allow CE if entering from the air)	
(b)	(i)	$n_g/n_w = 1.60/1.33 (= 1.20) \checkmark$ $1.20 (= \sin 90/\sin \theta_c) = 1/\sin \theta_c \checkmark$ $\theta_c = 56.2^\circ \checkmark$ $54^\circ$ is less than $\theta_c$ so reflection is not total $\checkmark$ [alternative ${}_a n_w = \sin \theta_a / \sin \theta_w; {}_a n_w = n_w/n_a \checkmark$ $1.60 \sin 54^\circ = 1.33 \sin \theta_w \checkmark$ $\theta_w = 76.7^\circ \checkmark$ refraction is possible so reflection is not total $\checkmark$ ]	6
	(ii)	ray refracting away from normal at the vertical block – water boundary $\checkmark$ ray refracting towards normal at the water – base boundary $\checkmark$	
<b>Total</b>			<b>11</b>

Question 5			
(a)		d and u $\checkmark$ (diagram may show the other two unaffected d and u quarks) $W^-$ with slope or arrow $\checkmark$ $e^-$ and $\bar{\nu}_{(e)}$ $\checkmark$	3
(b)		charge; baryon number; lepton number $\checkmark$ property 1Q: $1 - 1 = 0 + 0 \checkmark$ property 2B: $1 + 0 = 1 + 0 \checkmark$ property 3L: $0 + 1 = 0 + 1 \checkmark$ in any order	4
<b>Total</b>			<b>7</b>

<b>Question 6</b>		
(a)	referring to particle behaviour and wave behaviour ✓ example of particle behaviour such as the deflection of the electron in an electric or magnetic field ✓ example of wave behaviour such as the diffraction of electrons by a crystal ✓	<b>3</b>
(b)	(i) $v = h/m \lambda$ ✓ $= 6.63 \times 10^{-34} / (9.11 \times 10^{-31} \times 2.30 \times 10^{-10})$ ✓ $= 3.16 \times 10^6 \text{ m s}^{-1}$ ✓ (ii) $(m_\tau = m_e v_e / v_\tau) = 9.11 \times 10^{-31} \times 3.16 \times 10^6 / 911$ ✓ $m_\tau = 3.16 \times 10^{-27} \text{ kg}$ ✓	<b>5</b>
<b>Total</b>		<b>8</b>

Quality of Written Communication Q3 (a) and/or Q6 (a)	<b>2</b>
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